

t-3,*t*-5-Dimethyl-*r*-2,*c*-6-diphenyl-piperidin-4-one

S. Balamurugan,^a A. Thiruvalluvar,^{a*} A. Manimekalai^b and J. Jayabharathi^b

^aPG Research Department of Physics, Rajah Serfoji Government College (Autonomous), Thanjavur 613 005, Tamilnadu, India, and ^bDepartment of Chemistry, Annamalai University, Annamalai Nagar 608 002, Tamilnadu, India
Correspondence e-mail: athiru@vsnl.net

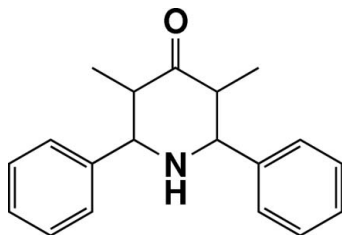
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Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.043; wR factor = 0.127; data-to-parameter ratio = 18.6.

In the title molecule, $\text{C}_{19}\text{H}_{21}\text{NO}$, the piperidine ring adopts a chair conformation. Two phenyl rings and two methyl groups, attached to the piperidine ring at positions 2, 6, 3 and 5, respectively, occupy equatorial positions. The dihedral angle between the two phenyl rings is 57.1 (1)°. Molecules are linked by $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds.

Related literature

For related literature, see: Noller & Baliah (1948); Hasan *et al.* (1985); Balamurugan *et al.* (2006, 2007); Thiruvalluvar, Balamurugan, Jayabharathi, Manimekalai & Rajarajan (2007); Thiruvalluvar, Balamurugan, Jayabharathi & Manimekalai (2007).



Experimental

Crystal data

$\text{C}_{19}\text{H}_{21}\text{NO}$
 $M_r = 279.37$
Triclinic, $P\bar{1}$
 $a = 7.1392$ (3) Å

$b = 10.5811$ (5) Å
 $c = 11.6276$ (5) Å
 $\alpha = 102.320$ (2)°
 $\beta = 107.613$ (2)°

$\gamma = 101.735$ (2)°
 $V = 783.31$ (6) Å³
 $Z = 2$
Mo $K\alpha$ radiation

$\mu = 0.07$ mm⁻¹
 $T = 298$ (2) K
 $0.30 \times 0.22 \times 0.18$ mm

Data collection

Bruker SMART APEXII diffractometer
Absorption correction: multi-scan (SADABS; Bruker, 2004)
 $T_{\min} = 0.854$, $T_{\max} = 0.987$

17560 measured reflections
3606 independent reflections
2507 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.022$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.043$
 $wR(F^2) = 0.127$
 $S = 1.05$
3606 reflections
194 parameters

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.18$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.15$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1}\cdots\text{O4}^i$	0.90 (2)	2.43 (2)	3.3109 (17)	170.2 (15)

Symmetry code: (i) $x - 1, y, z$.

Data collection: *APEX2* (Bruker, 2004); cell refinement: *APEX2*; data reduction: *SAINT-NT* (Bruker, 2004); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *PLATON* (Spek, 2003).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: IS2193).

References

- Balamurugan, S., Thiruvalluvar, A., Manimekalai, A., Selvaraju, K. & Maruthavanan, T. (2006). *Acta Cryst.* **E62**, o2005–o2006.
Balamurugan, S., Thiruvalluvar, A., Manimekalai, A., Selvaraju, K. & Maruthavanan, T. (2007). *Acta Cryst.* **E63**, o789–o791.
Bruker (2004). *APEX2* (Version 1.22), *SAINT-NT* (Version 6.0) and *SADABS* (Version 2004/1). Bruker AXS Inc., Madison, Wisconsin, USA.
Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
Hasan, M. V., Arab, M., Pandiarajan, K., Sekar, R. & Marko, D. (1985). *Magn. Reson. Chem.* **23**, 292–295.
Noller, C. R. & Baliah, V. (1948). *J. Am. Chem. Soc.* **70**, 3853–3855.
Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
Thiruvalluvar, A., Balamurugan, S., Jayabharathi, J. & Manimekalai, A. (2007). *Acta Cryst.* **E63**, o2910.
Thiruvalluvar, A., Balamurugan, S., Jayabharathi, J., Manimekalai, A. & Rajarajan, G. (2007). *Acta Cryst.* **E63**, o2886.

supplementary materials

Acta Cryst. (2007). E63, o3504 [doi:10.1107/S1600536807034010]

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Comment

The conformation of the title compound was established by NMR spectroscopy by Hasan *et al.* (1985). Crystal structures of di-2-furylpiperidin-4-one derivatives have been reported, wherein the piperidine ring adopts a chair (Balamurugan *et al.*, 2006), a twist-boat (Balamurugan *et al.*, 2007) and a chair conformation (Thiruvalluvar, Balamurugan, Jayabharathi & Manimekalai, 2007). Thiruvalluvar, Balamurugan, Jayabharathi, Manimekalai & Rajarajan (2007) have reported the crystal structure of diphenylpiperidin-4-ol derivative, wherein the piperidine ring adopts a chair conformation.

In the title molecule, C₁₉H₂₁NO, the piperidine ring adopts a chair conformation (Fig. 1). Two phenyl rings and two methyl groups attached to the piperidine ring at the positions 2, 6, 3 and 5, respectively, have equatorial orientations. The dihedral angle between the two phenyl rings is 57.1 (1)°. Molecules are linked by an N1—H1···O4 (−1 + *x*, *y*, *z*) hydrogen bond (Fig. 2).

Experimental

The title compound was prepared by the known procedure (Noller & Baliah, 1948) and characterized using NMR techniques (Hasan *et al.*, 1985).

Refinement

Atom H1 at N1 was located in a difference Fourier map and refined isotropically. Remaining H atoms were positioned geometrically and allowed to ride on their parent atoms, with C—H = 0.93–0.98 Å and with $U_{\text{iso}}(\text{H}) = 1.2\text{--}1.5U_{\text{eq}}(\text{C})$.

Figures

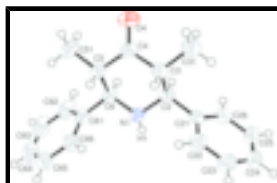


Fig. 1. The molecular structure of the title compound, showing the atom-numbering scheme and displacement ellipsoids drawn at the 50% probability level.

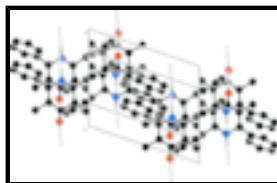


Fig. 2. The molecular packing of the title compound, viewed down the *b* axis, showing the hydrogen bonds (dashed lines).

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Crystal data

$C_{19}H_{21}NO$	$Z = 2$
$M_r = 279.37$	$F_{000} = 300$
Triclinic, $P\bar{1}$	$D_x = 1.184 \text{ Mg m}^{-3}$
Hall symbol: -P 1	Melting point: 405(1) K
$a = 7.1392 (3) \text{ \AA}$	Mo $K\alpha$ radiation
$b = 10.5811 (5) \text{ \AA}$	$\lambda = 0.71073 \text{ \AA}$
$c = 11.6276 (5) \text{ \AA}$	Cell parameters from 10924 reflections
$\alpha = 102.320 (2)^\circ$	$\theta = 1.9\text{--}27.8^\circ$
$\beta = 107.613 (2)^\circ$	$\mu = 0.07 \text{ mm}^{-1}$
$\gamma = 101.735 (2)^\circ$	$T = 298 (2) \text{ K}$
$V = 783.31 (6) \text{ \AA}^3$	Block, yellow
	$0.30 \times 0.22 \times 0.18 \text{ mm}$

Data collection

Bruker SMART APEXII diffractometer	3606 independent reflections
Radiation source: fine-focus sealed tube	2507 reflections with $I > 2\sigma(I)$
Monochromator: graphite	$R_{\text{int}} = 0.022$
$T = 298(2) \text{ K}$	$\theta_{\text{max}} = 27.8^\circ$
φ and ω scans	$\theta_{\text{min}} = 1.9^\circ$
Absorption correction: multi-scan (SADABS; Bruker, 2004)	$h = -9 \rightarrow 9$
$T_{\text{min}} = 0.854, T_{\text{max}} = 0.987$	$k = -13 \rightarrow 13$
17560 measured reflections	$l = -15 \rightarrow 15$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.043$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.127$	$w = 1/[\sigma^2(F_o^2) + (0.0551P)^2 + 0.1535P]$
$S = 1.05$	where $P = (F_o^2 + 2F_c^2)/3$
3606 reflections	$(\Delta/\sigma)_{\text{max}} < 0.001$
194 parameters	$\Delta\rho_{\text{max}} = 0.18 \text{ e \AA}^{-3}$
Primary atom site location: structure-invariant direct methods	$\Delta\rho_{\text{min}} = -0.15 \text{ e \AA}^{-3}$
	Extinction correction: none

Special details

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.43830 (15)	0.75505 (11)	0.24410 (10)	0.0382 (3)
H1	0.331 (3)	0.7840 (16)	0.2495 (14)	0.058 (4)*
C2	0.49854 (18)	0.79381 (13)	0.14409 (12)	0.0377 (3)
H2	0.5555	0.8923	0.1702	0.045*
C3	0.6675 (2)	0.72851 (15)	0.12688 (13)	0.0452 (3)
H3	0.6095	0.6303	0.1032	0.054*
C4	0.8407 (2)	0.77112 (15)	0.25275 (14)	0.0478 (3)
O4	1.01600 (16)	0.82417 (15)	0.26561 (12)	0.0783 (4)
C5	0.78155 (19)	0.74549 (15)	0.36128 (13)	0.0469 (3)
H5	0.7275	0.6475	0.3422	0.056*
C6	0.60440 (18)	0.80756 (13)	0.36724 (12)	0.0390 (3)
H6	0.6554	0.9058	0.3869	0.047*
C21	0.31407 (18)	0.75191 (13)	0.02317 (12)	0.0385 (3)
C22	0.2010 (2)	0.61779 (15)	-0.03430 (14)	0.0500 (3)
H22	0.2367	0.5525	0.0031	0.060*
C23	0.0361 (2)	0.57985 (17)	-0.14633 (14)	0.0590 (4)
H23	-0.0375	0.4892	-0.1845	0.071*
C24	-0.0200 (2)	0.67568 (18)	-0.20191 (14)	0.0589 (4)
H24	-0.1312	0.6500	-0.2775	0.071*
C25	0.0882 (2)	0.80852 (17)	-0.14564 (14)	0.0564 (4)
H25	0.0501	0.8736	-0.1826	0.068*
C26	0.2547 (2)	0.84655 (15)	-0.03351 (13)	0.0464 (3)
H26	0.3276	0.9373	0.0042	0.056*
C31	0.7402 (3)	0.7603 (2)	0.02373 (16)	0.0680 (5)
H31A	0.6255	0.7314	-0.0546	0.102*
H31B	0.8396	0.7138	0.0148	0.102*
H31C	0.8016	0.8559	0.0456	0.102*
C51	0.9625 (2)	0.7943 (2)	0.48516 (16)	0.0728 (5)
H51A	0.9169	0.7753	0.5510	0.109*
H51B	1.0219	0.8899	0.5054	0.109*
H51C	1.0634	0.7487	0.4776	0.109*
C61	0.53009 (19)	0.77581 (14)	0.46923 (12)	0.0425 (3)
C62	0.5871 (2)	0.87132 (18)	0.58451 (14)	0.0602 (4)
H62	0.6657	0.9587	0.5985	0.072*

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C63	0.5270 (3)	0.8370 (2)	0.67992 (15)	0.0767 (6)
H63	0.5681	0.9014	0.7578	0.092*
C64	0.4088 (3)	0.7101 (3)	0.66034 (17)	0.0745 (6)
H64	0.3681	0.6882	0.7241	0.089*
C65	0.3509 (3)	0.6158 (2)	0.54676 (17)	0.0659 (5)
H65	0.2700	0.5292	0.5330	0.079*
C66	0.4113 (2)	0.64769 (16)	0.45176 (14)	0.0511 (4)
H66	0.3713	0.5819	0.3749	0.061*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O4	0.0375 (5)	0.1211 (11)	0.0795 (9)	0.0161 (6)	0.0247 (5)	0.0370 (8)
N1	0.0302 (5)	0.0506 (7)	0.0349 (6)	0.0144 (5)	0.0118 (4)	0.0120 (5)
C2	0.0337 (6)	0.0413 (7)	0.0395 (7)	0.0125 (5)	0.0138 (5)	0.0123 (5)
C3	0.0411 (7)	0.0547 (8)	0.0489 (8)	0.0204 (6)	0.0228 (6)	0.0173 (7)
C4	0.0362 (6)	0.0569 (8)	0.0603 (9)	0.0223 (6)	0.0215 (6)	0.0226 (7)
C5	0.0372 (7)	0.0601 (9)	0.0509 (8)	0.0228 (6)	0.0156 (6)	0.0232 (7)
C6	0.0324 (6)	0.0427 (7)	0.0390 (7)	0.0129 (5)	0.0089 (5)	0.0102 (6)
C21	0.0359 (6)	0.0476 (8)	0.0349 (7)	0.0128 (5)	0.0163 (5)	0.0124 (6)
C22	0.0496 (8)	0.0481 (8)	0.0492 (8)	0.0123 (6)	0.0152 (6)	0.0134 (7)
C23	0.0536 (8)	0.0592 (9)	0.0485 (9)	0.0039 (7)	0.0130 (7)	0.0040 (7)
C24	0.0473 (8)	0.0830 (12)	0.0363 (8)	0.0090 (8)	0.0094 (6)	0.0150 (8)
C25	0.0546 (8)	0.0723 (11)	0.0461 (8)	0.0194 (8)	0.0144 (7)	0.0290 (8)
C26	0.0451 (7)	0.0510 (8)	0.0436 (8)	0.0123 (6)	0.0153 (6)	0.0176 (6)
C31	0.0589 (9)	0.1070 (14)	0.0626 (10)	0.0384 (9)	0.0378 (8)	0.0371 (10)
C51	0.0436 (8)	0.1164 (15)	0.0646 (11)	0.0352 (9)	0.0114 (7)	0.0395 (10)
C61	0.0376 (6)	0.0565 (9)	0.0338 (7)	0.0231 (6)	0.0090 (5)	0.0108 (6)
C62	0.0613 (9)	0.0684 (10)	0.0442 (9)	0.0308 (8)	0.0101 (7)	0.0045 (8)
C63	0.0870 (13)	0.1156 (17)	0.0349 (9)	0.0604 (13)	0.0194 (8)	0.0113 (10)
C64	0.0760 (12)	0.1259 (18)	0.0549 (11)	0.0593 (12)	0.0361 (9)	0.0466 (12)
C65	0.0635 (10)	0.0901 (13)	0.0654 (11)	0.0327 (9)	0.0328 (8)	0.0417 (10)
C66	0.0486 (8)	0.0627 (9)	0.0454 (8)	0.0186 (7)	0.0189 (6)	0.0175 (7)

Geometric parameters (\AA , $^\circ$)

O4—C4	1.212 (2)	C64—C65	1.360 (3)
N1—C2	1.4628 (17)	C65—C66	1.381 (3)
N1—C6	1.4595 (17)	C2—H2	0.9800
N1—H1	0.90 (2)	C3—H3	0.9800
C2—C3	1.549 (2)	C5—H5	0.9800
C2—C21	1.5112 (18)	C6—H6	0.9800
C3—C31	1.516 (2)	C22—H22	0.9300
C3—C4	1.507 (2)	C23—H23	0.9300
C4—C5	1.505 (2)	C24—H24	0.9300
C5—C51	1.515 (2)	C25—H25	0.9300
C5—C6	1.552 (2)	C26—H26	0.9300
C6—C61	1.5096 (19)	C31—H31A	0.9600
C21—C22	1.384 (2)	C31—H31B	0.9600

C21—C26	1.379 (2)	C31—H31C	0.9600
C22—C23	1.379 (2)	C51—H51A	0.9600
C23—C24	1.376 (3)	C51—H51B	0.9600
C24—C25	1.364 (3)	C51—H51C	0.9600
C25—C26	1.385 (2)	C62—H62	0.9300
C61—C66	1.382 (2)	C63—H63	0.9300
C61—C62	1.381 (2)	C64—H64	0.9300
C62—C63	1.394 (3)	C65—H65	0.9300
C63—C64	1.364 (4)	C66—H66	0.9300
O4…H1 ⁱ	2.43 (2)	H5…H65 ^{iv}	2.6000
O4…H31B	2.6800	H6…H2	2.3600
O4…H31C	2.6800	H6…H62	2.3800
O4…H51B	2.7100	H6…H51B ⁱⁱ	2.5700
O4…H51C	2.7000	H22…N1	2.8700
O4…H62 ⁱⁱ	2.6600	H22…C3	2.9700
N1…H22	2.8700	H22…H3	2.4400
N1…H66	2.6800	H23…H66 ^v	2.5500
C24…C31 ⁱⁱⁱ	3.599 (3)	H24…H66 ^v	2.4800
C31…C24 ⁱ	3.599 (3)	H26…H2	2.3300
C51…C62	3.398 (3)	H26…H26 ^{vii}	2.5900
C62…C51	3.398 (3)	H31A…C21	2.6700
C65…C66 ^{iv}	3.547 (3)	H31A…H64 ^{viii}	2.5500
C66…C65 ^{iv}	3.547 (3)	H31B…O4	2.6800
C3…H22	2.9700	H31B…C23 ⁱ	2.9600
C21…H31A	2.6700	H31B…C24 ⁱ	2.9600
C22…H3	2.8300	H31C…O4	2.6800
C23…H31B ⁱⁱⁱ	2.9600	H51A…C61	2.6400
C24…H66 ^v	3.1000	H51A…C62	2.8400
C24…H31B ⁱⁱⁱ	2.9600	H51B…O4	2.7100
C61…H51A	2.6400	H51B…H6 ⁱⁱ	2.5700
C62…H51A	2.8400	H51B…H51B ⁱⁱ	2.4400
C64…H51C ⁱⁱⁱ	2.8900	H51C…O4	2.7000
C65…H51C ⁱⁱⁱ	2.7400	H51C…C64 ⁱ	2.8900
C66…H5	2.9100	H51C…C65 ⁱ	2.7400
C66…H51C ⁱⁱⁱ	2.9700	H51C…C66 ⁱ	2.9700
C66…H1	2.985 (16)	H62…H6	2.3800
H1…O4 ⁱⁱⁱ	2.43 (2)	H62…O4 ⁱⁱ	2.6600
H1…C66	2.985 (16)	H63…H2 ^{vi}	2.5900
H2…H6	2.3600	H64…H31A ^{ix}	2.5500
H2…H26	2.3300	H65…H5 ^{iv}	2.6000
H2…H63 ^{vi}	2.5900	H66…N1	2.6800
H3…C22	2.8300	H66…C24 ^v	3.1000
H3…H22	2.4400	H66…H23 ^v	2.5500

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H5...C66	2.9100	H66...H24 ^v	2.4800
C2—N1—C6	112.91 (11)	C31—C3—H3	108.00
C2—N1—H1	110.0 (10)	C4—C5—H5	107.00
C6—N1—H1	109.1 (10)	C6—C5—H5	107.00
C3—C2—C21	111.82 (11)	C51—C5—H5	107.00
N1—C2—C21	110.33 (11)	N1—C6—H6	109.00
N1—C2—C3	108.65 (11)	C5—C6—H6	109.00
C2—C3—C31	113.35 (14)	C61—C6—H6	109.00
C4—C3—C31	112.24 (14)	C21—C22—H22	120.00
C2—C3—C4	108.12 (12)	C23—C22—H22	120.00
C3—C4—C5	115.81 (13)	C22—C23—H23	120.00
O4—C4—C3	121.85 (14)	C24—C23—H23	120.00
O4—C4—C5	122.34 (14)	C23—C24—H24	120.00
C6—C5—C51	113.06 (12)	C25—C24—H24	120.00
C4—C5—C6	108.70 (12)	C24—C25—H25	120.00
C4—C5—C51	112.75 (13)	C26—C25—H25	120.00
N1—C6—C5	108.99 (11)	C21—C26—H26	119.00
N1—C6—C61	111.00 (11)	C25—C26—H26	119.00
C5—C6—C61	110.51 (11)	C3—C31—H31A	109.00
C2—C21—C22	121.10 (13)	C3—C31—H31B	109.00
C22—C21—C26	118.14 (13)	C3—C31—H31C	109.00
C2—C21—C26	120.76 (13)	H31A—C31—H31B	109.00
C21—C22—C23	120.80 (15)	H31A—C31—H31C	109.00
C22—C23—C24	120.21 (16)	H31B—C31—H31C	109.00
C23—C24—C25	119.72 (14)	C5—C51—H51A	109.00
C24—C25—C26	120.07 (16)	C5—C51—H51B	109.00
C21—C26—C25	121.05 (15)	C5—C51—H51C	109.00
C6—C61—C62	121.21 (14)	H51A—C51—H51B	109.00
C62—C61—C66	118.15 (14)	H51A—C51—H51C	109.00
C6—C61—C66	120.56 (12)	H51B—C51—H51C	109.00
C61—C62—C63	120.10 (17)	C61—C62—H62	120.00
C62—C63—C64	120.74 (16)	C63—C62—H62	120.00
C63—C64—C65	119.5 (2)	C62—C63—H63	120.00
C64—C65—C66	120.5 (2)	C64—C63—H63	120.00
C61—C66—C65	121.02 (15)	C63—C64—H64	120.00
N1—C2—H2	109.00	C65—C64—H64	120.00
C3—C2—H2	109.00	C64—C65—H65	120.00
C21—C2—H2	109.00	C66—C65—H65	120.00
C2—C3—H3	108.00	C61—C66—H66	120.00
C4—C3—H3	108.00	C65—C66—H66	119.00
C6—N1—C2—C3	64.29 (14)	C51—C5—C6—N1	179.84 (13)
C6—N1—C2—C21	-172.79 (11)	C51—C5—C6—C61	-57.92 (17)
C2—N1—C6—C5	-63.18 (14)	N1—C6—C61—C62	-136.85 (15)
C2—N1—C6—C61	174.88 (11)	N1—C6—C61—C66	46.50 (18)
N1—C2—C3—C4	-55.70 (15)	C5—C6—C61—C62	102.10 (17)
N1—C2—C3—C31	179.19 (13)	C5—C6—C61—C66	-74.55 (17)
C21—C2—C3—C4	-177.71 (12)	C2—C21—C22—C23	-178.30 (13)
C21—C2—C3—C31	57.18 (17)	C26—C21—C22—C23	1.2 (2)

N1—C2—C21—C22	-59.92 (17)	C2—C21—C26—C25	178.73 (13)
N1—C2—C21—C26	120.57 (14)	C22—C21—C26—C25	-0.8 (2)
C3—C2—C21—C22	61.12 (17)	C21—C22—C23—C24	-0.8 (2)
C3—C2—C21—C26	-118.38 (15)	C22—C23—C24—C25	0.0 (2)
C2—C3—C4—O4	-125.93 (17)	C23—C24—C25—C26	0.5 (2)
C2—C3—C4—C5	53.21 (17)	C24—C25—C26—C21	0.0 (2)
C31—C3—C4—O4	-0.2 (2)	C6—C61—C62—C63	-175.98 (17)
C31—C3—C4—C5	178.97 (14)	C66—C61—C62—C63	0.8 (3)
O4—C4—C5—C6	126.84 (17)	C6—C61—C66—C65	176.81 (16)
O4—C4—C5—C51	0.7 (2)	C62—C61—C66—C65	0.1 (3)
C3—C4—C5—C6	-52.29 (17)	C61—C62—C63—C64	-1.1 (3)
C3—C4—C5—C51	-178.47 (14)	C62—C63—C64—C65	0.7 (3)
C4—C5—C6—N1	53.83 (15)	C63—C64—C65—C66	0.2 (3)
C4—C5—C6—C61	176.07 (12)	C64—C65—C66—C61	-0.5 (3)

Symmetry codes: (i) $x+1, y, z$; (ii) $-x+2, -y+2, -z+1$; (iii) $x-1, y, z$; (iv) $-x+1, -y+1, -z+1$; (v) $-x, -y+1, -z$; (vi) $-x+1, -y+2, -z+1$; (vii) $-x+1, -y+2, -z$; (viii) $x, y, z-1$; (ix) $x, y, z+1$.

Hydrogen-bond geometry ($\text{\AA}, ^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1 \cdots O4 ⁱⁱⁱ	0.90 (2)	2.43 (2)	3.3109 (17)	170.2 (15)

Symmetry codes: (iii) $x-1, y, z$.

Fig. 1

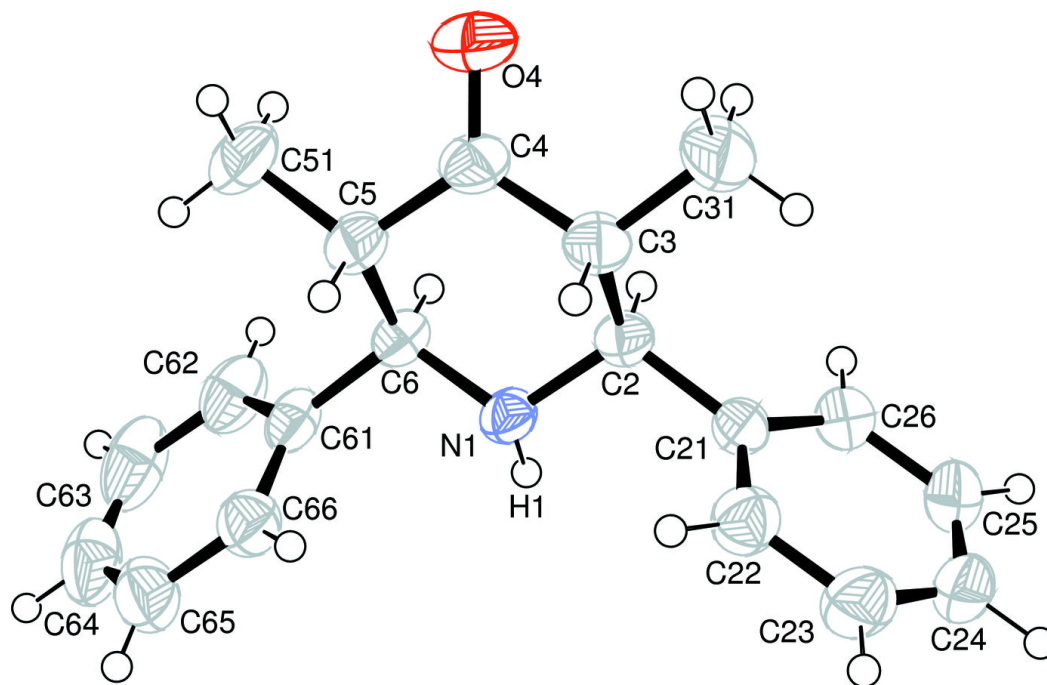


Fig. 2

